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Key indicators

Single-crystal X-ray study T = 299 KMean $\sigma(C-C) = 0.003 \text{ Å}$ R factor = 0.035 wR factor = 0.101 Data-to-parameter ratio = 10.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

In the crystal structure of the title compound, $C_8H_7ClN_2O_3$, the molecules are linked into chains by $N-H \cdots O$ hydrogen bonds.

2-Chloro-N-(4-nitrophenyl)acetamide

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Comment

As part of a study on the systematization of the crystal structures of acetanilides (Gowda *et al.*, 2006; Gowda, Kozisek *et al.*, 2007; Gowda, Foro & Fuess, 2007), the structure of 2-chloro-*N*-(4-nitrophenyl)acetamide (4NP2CA), (I), has been determined to explore the effects of polar substituent groups on the structure of *N*-aromatic amides. In 4NP2CA, the amide H atom is involved in intermolecular hydrogen bonding with the O atom (Table 1).



Experimental

The title compound, (I), was prepared according to a literature method (Gowda & Weiss, 1994). The purity of the compound was checked by determining its melting point. It was characteried by recording its IR and NQR spectra (Gowda & Weiss, 1994). Single crystals of the title compound were obtained from an ethanol solution.



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Figure 1

The molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level.

organic papers

Data collection

Enraf-Nonius CAD-4 diffractometer Absorption correction: ψ scan (North *et al.*, 1968) $T_{\rm min} = 0.353, T_{\rm max} = 0.834$ 5404 measured reflections

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.035$ $wR(F^2) = 0.101$ S = 1.041610 reflections 149 parameters 1 restraint

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots \mathbf{A}$
$N5-H5N\cdots O4^{i}$	0.845 (10)	2.040 (12)	2.840 (2)	158 (2)
Symmetry code: (i)	$x, -y + \frac{1}{2}, z - \frac{1}{2}.$			

1610 independent reflections

3 standard reflections

frequency: 120 min

intensity decay: 3.5%

H atoms treated by a mixture of independent and constrained

 $R_{\rm int} = 0.038$

refinement $\Delta \rho_{\text{max}} = 0.21 \text{ e } \text{\AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.23 \text{ e} \text{ Å}^{-3}$

1294 reflections with $I > 2\sigma(I)$

H atoms bonded to C were positioned geometrically and treated as riding, with C–H = 0.93 (aromatic) or 0.97 Å (CH₂), and with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$. The coordinates of the H atom bonded to nitrogen were refined with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm N})$.

Data collection: *CAD-4-PC* (Nonius, 1996); cell refinement: *CAD-4-PC*; data reduction: *REDU4* (Stoe & Cie, 1987); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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Figure 2

Partial packing diagram of the title compound, viewed in the bc plane. Hydrogen bonds are shown as dashed lines.

References

Gowda, B. T., Foro, S. & Fuess, H. (2007). Acta Cryst. E63, 01975-01976.

- Gowda, B. T., Kozisek, J. & Fuess, H. (2006). Z. Naturforsch. Teil A, 61, 588– 594.
- Gowda, B. T., Kozisek, J., Svoboda, I. & Fuess, H. (2007). Z. Naturforsch. Teil A, 62, 91–100.
- Gowda, B. T. & Weiss, A. (1994). Z. Naturforsch. Teil A, 49, 695-702.
- Nonius (1996). CAD-4-PC. Nonius, Delft, The Netherlands.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351–359.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
- Stoe & Cie (1987). REDU4. Stoe & Cie GmbH, Darmstadt, Germany.